

Correlating Composition, Morphology and Performance Metrologies Performed on Organic Photovoltaics: A Combinatorial Challenge

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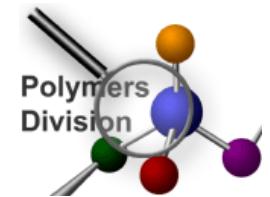
**National Institute of
Standards and Technology**
Technology Administration
U.S. Department of Commerce



NIST Combinatorial Methods Center

Meeting No. 14: A DOE-EERE/NIST Joint Workshop on Combinatorial Materials Science for Applications in Energy

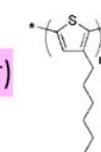
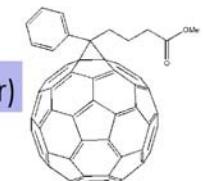
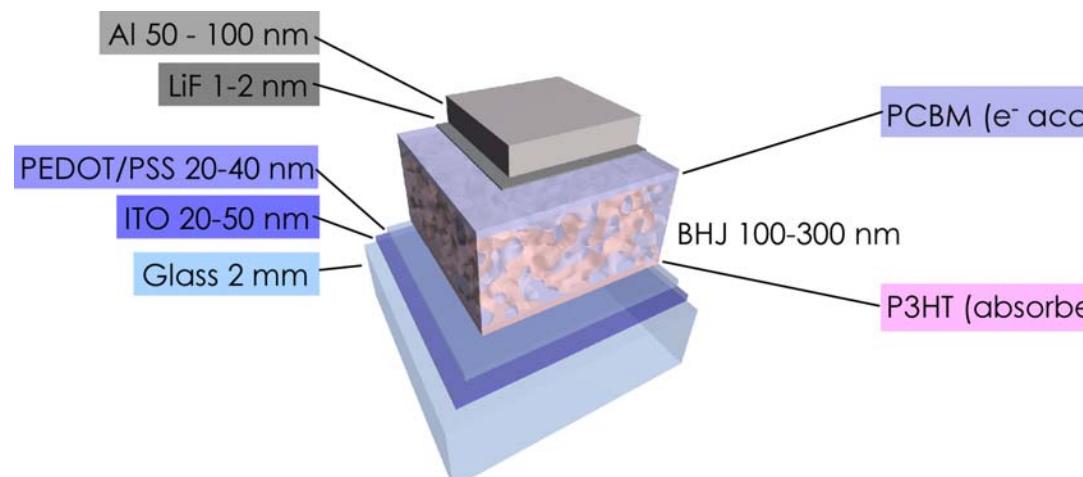
Next-Generation Photovoltaics: The Big Deal About a Small Thing



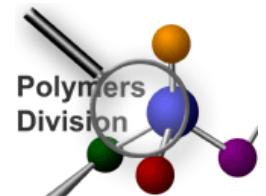
Next-gen photovoltaics: printable & flexible thin-film, organic, and hybrid solar cells, which rely on complex nanostructured morphologies of multiple components.



The Promise: low-cost, large-area fabrication (printing, doctor-blading, painting), light weight, flexible/ foldable/ rollable substrates.



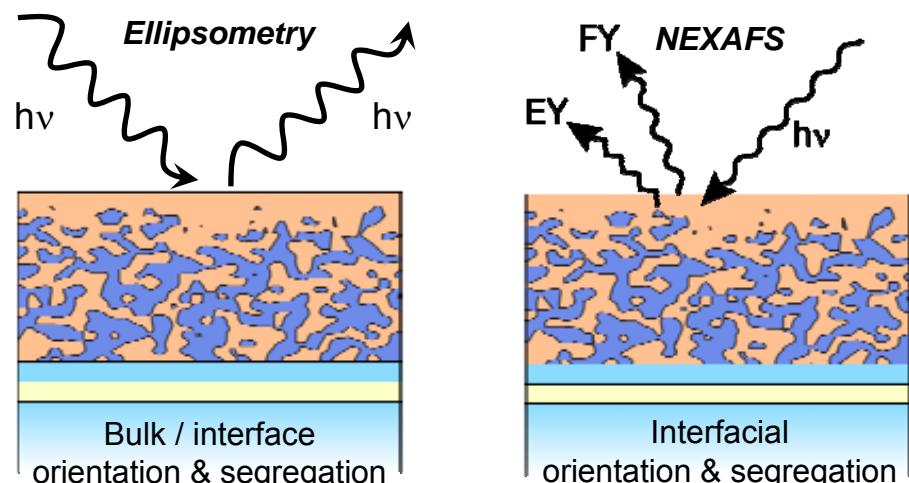
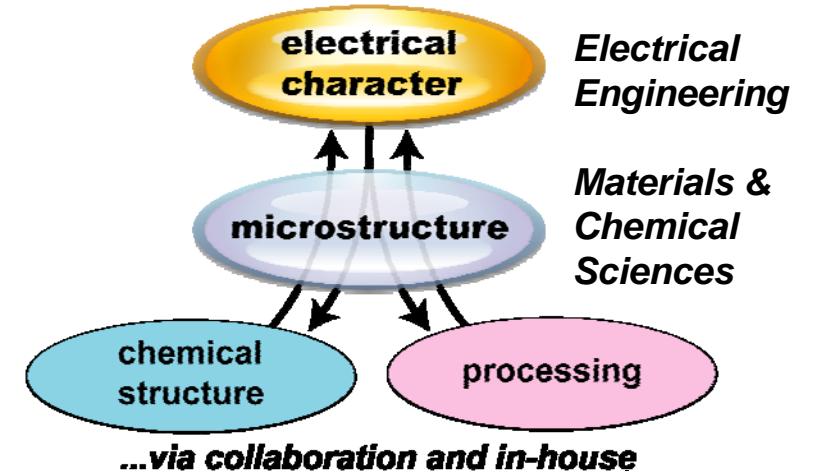
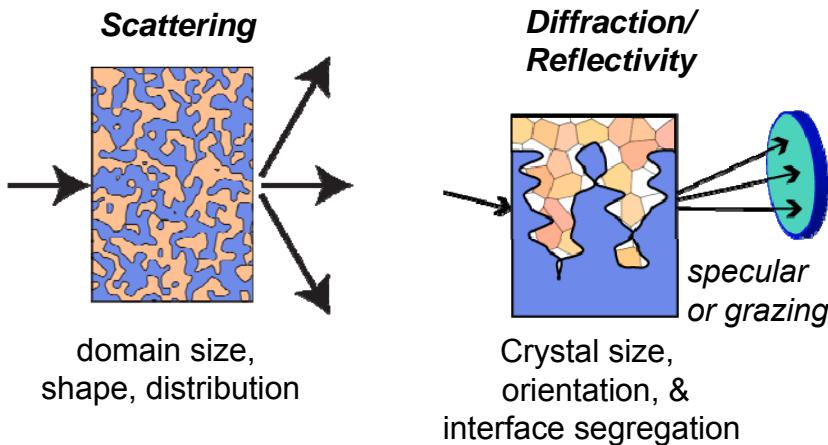
The Reality: low efficiencies related to uncontrolled and poorly understood interface composition and active layer morphology, poor solar spectrum overlap, band-gap mismatches and short useful device lifetimes.

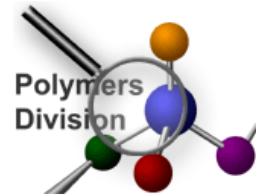


NIST Organic Photovoltaics Program

Metrology to Enable the Realization of Organic Electronics (Started in '05)

- to “provide the integrated measurement and standards tools needed to accelerate progress in organic electronics”
- Satisfies a ubiquitous need expressed by organic electronics technology developers.





NIST Organic Electronics & Photovoltaics Team

Polymer Division

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Surface and Microanalysis Division

Lee Richter

Semiconductor Electronics Division

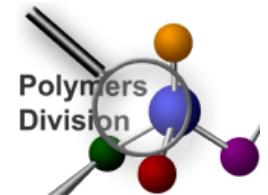
David Gundlach

Calvin Chan

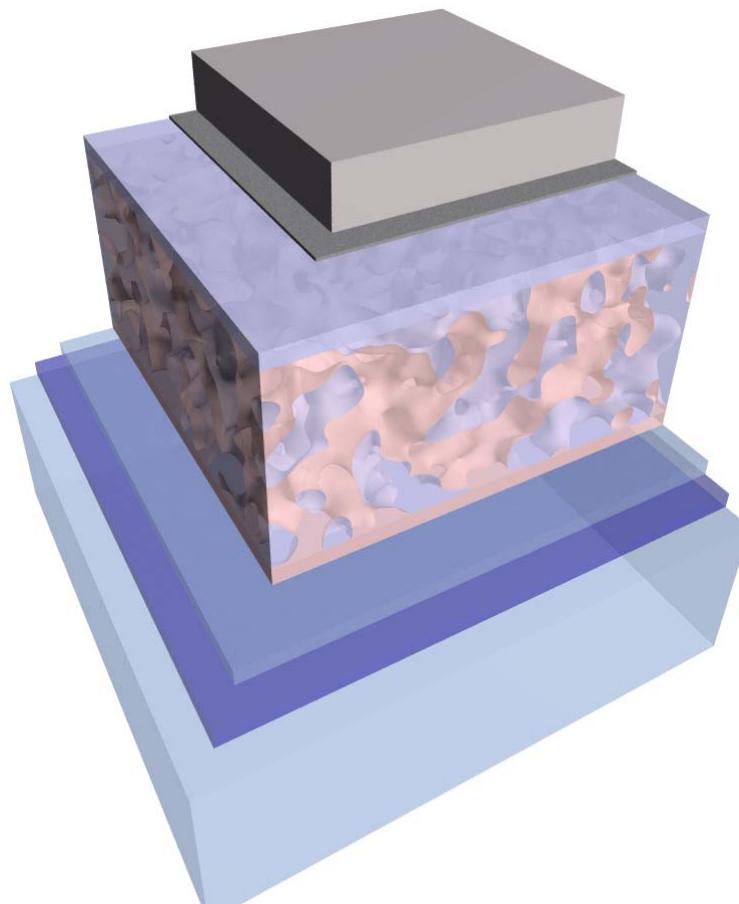
Oana Jurchescu

Behrang Hamadani*





Developing bulk heterojunction structural model



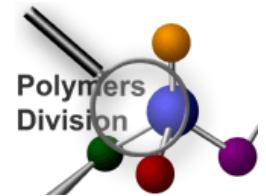
Molecular orientation

- Light absorbance (by P3HT)
- Transport anisotropy
- Interpretation of model-dependent methods

Vertical composition gradient

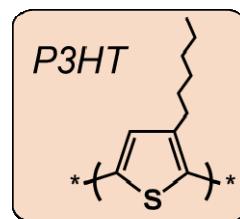
- Relates to series resistance, could affect device characteristics.
- Necessary to know this to interpret measurements of domain size and shape.

Surface segregation control by SAMs

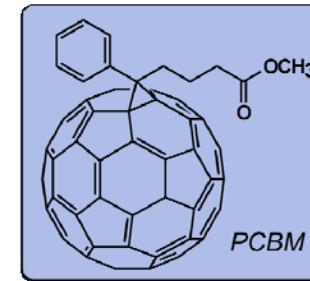


Substrates of different surface energy

- Vertical composition profile is likely to be influenced by substrate chemistry
- Solution-deposited SAMs provide model substrates
- In “real” OPV devices, the surface energy of the HIL influences the composition profile

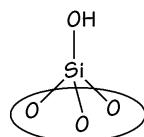


$$\gamma = 27 \text{ mN/m}^2$$

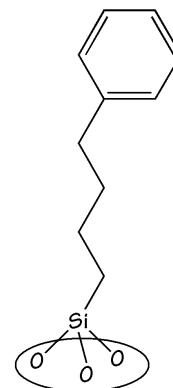


$$\gamma = 38 \text{ mN/m}^2$$

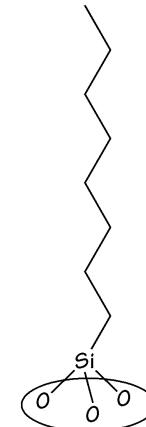
*Model surface
for PEDOT:PSS*



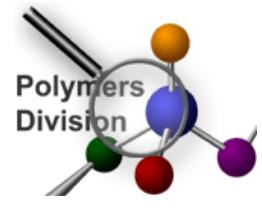
$$\gamma = 77 \text{ mN/m}^2$$



$$\gamma = 26 \text{ mN/m}^2$$



$$\gamma = 20 \text{ mN/m}^2$$



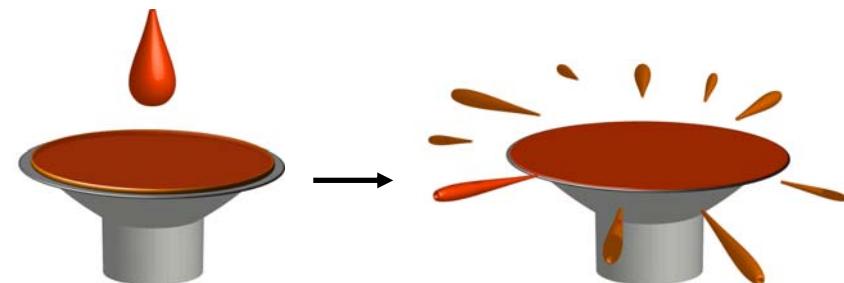
Device Fabrication

Bulk heterojunction (BHJ) Recipe:

- 30 mg/mL (total solids) in o-dichlorobenzene, 60 °C
- 1:1 weight ratio of P3HT:PCBM
- 500 rpm, 60 seconds, A = 001
- “As Cast” = solvent annealed[†]
- “Annealed” = 140 °C, 40 min

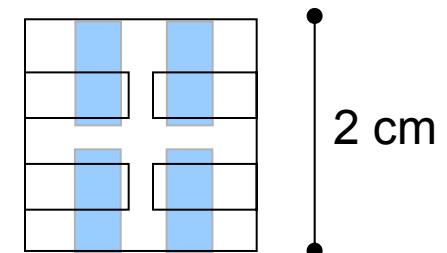
Substrate Preparation:

- Sonication 10 min in Acetone and IPA
- UV Ozone for 10 minutes
- Water rinse and N₂ Dry



SAMs

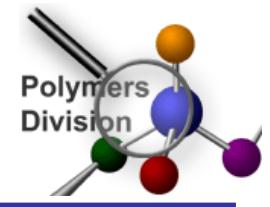
- 2 mM OTS in Hexadecane overnight
- Sonication in CHCl₃, IPA, and H₂O followed by 10 min at 150 °C
- 10 mM PBTS in toluene overnight [‡]
- Sonication in toluene, acetone, and IPA followed by 10 min at 150 °C



[†] Li, G. et al. *Adv. Funct. Mater.* **2007**, 17 (10), 1636-1644

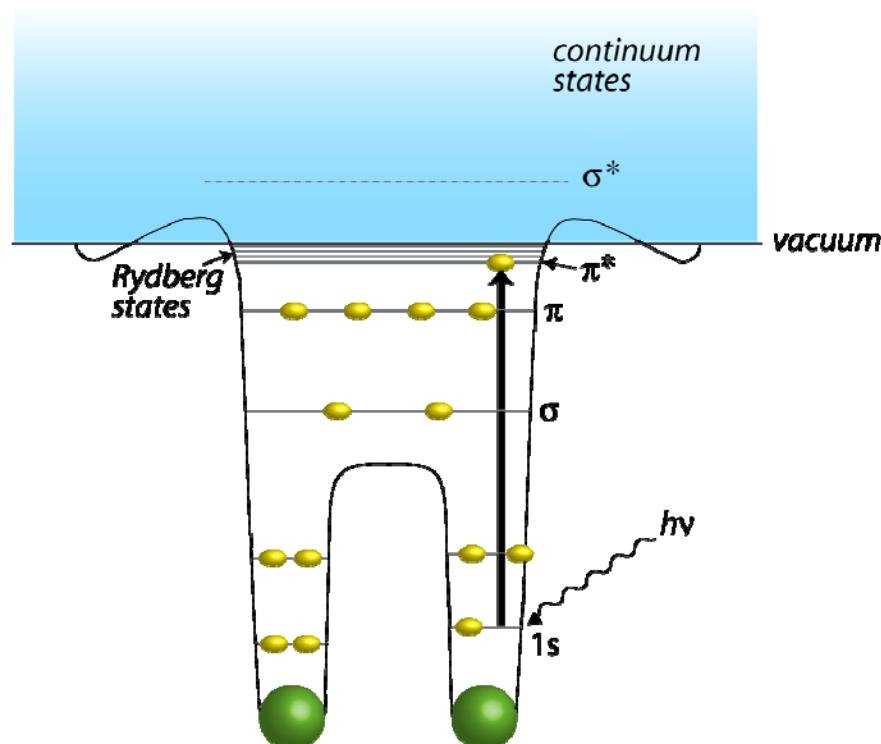
[‡] Kumaki, D. et. Al. *Appl. Phys. Lett.* **2007**, 90, 133511

Near-Edge X-ray Absorption Fine Structure: (NEXAFS) Spectroscopy



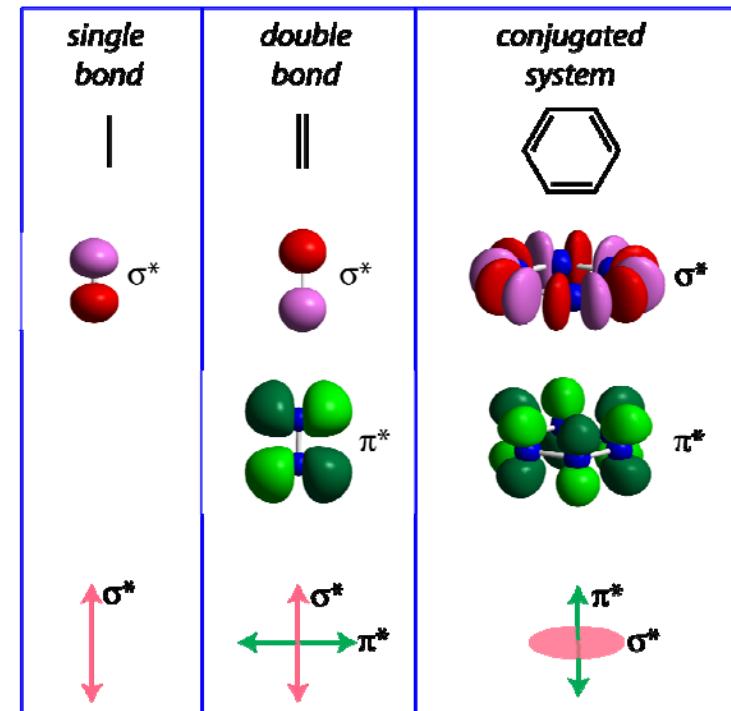
Resonant bound-state excitations

- Atomic electrons excited to unfilled molecular orbitals
- Element specific, and bond-sensitive
- Absorbance measured by Auger or fluorescence.

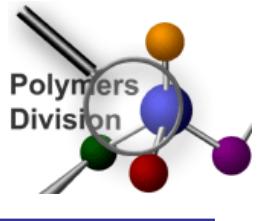


Resonance orientations

- Defined by end state orbital
- Can be decomposed into vector and plane representations
- The π^* resonance of organic semiconductors is especially useful because it is a single vector for any conjugated plane.



The Nuts and Bolts of NEXAFS Measurements: NIST/Dow Beamline U7A at the NSLS



Strengths for Organic Electronics:

- Detects C, N, O, & F bonds.
- High sensitivity to π bonding.
- Directly measures orientation.
- Depth sensitive.

Does NOT measure :

- Crystal packing style, prevalence, size, shape.
- HOMO, bandgaps.
- Secondary chemical interactions (e.g. vib / rot structure).

How to apply NEXAFS to P3HT & pentacene:

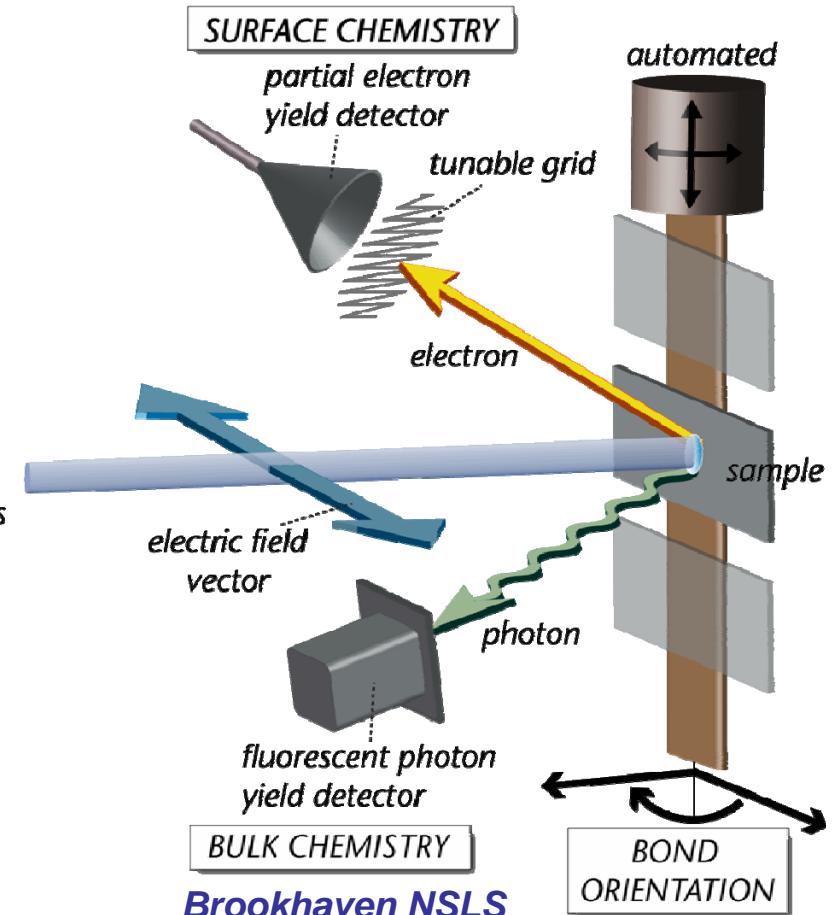
- Proc. SPIE OFET IV 2005 p54
- In "Organic Field Effect Transistors" eds. Locklin & Bao 2007

P3HT:

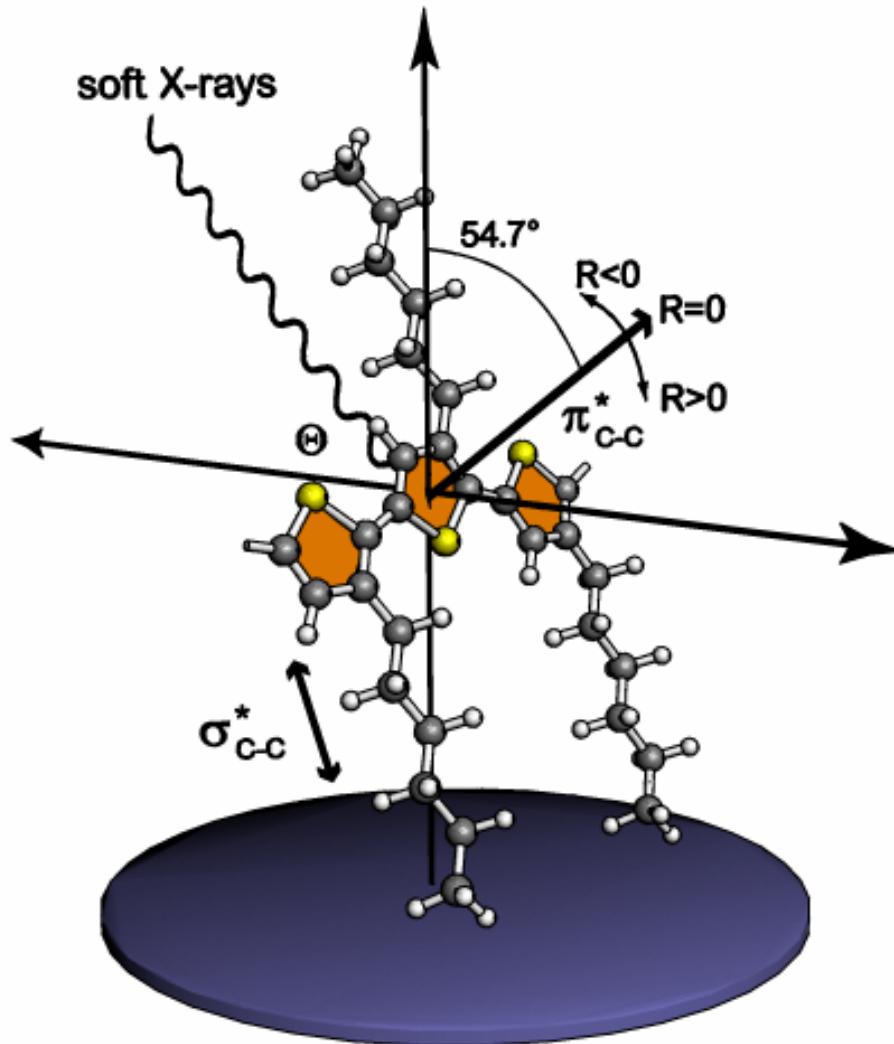
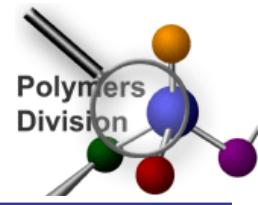
- Chem. Mater. 2005 p5610 (spin speed)
- Langmuir 2007 p834 (liquid side chains)

Oligothiophenes (w/ J.M.J. Fréchet & V. Subramanian):

- Adv. Mater. 2005 p2340
- Chem. Mater. 2006 p6033
- J. Phys. Chem. B. 2006 p10654



NEXAFS Example: P3HT



Resonance Intensity:

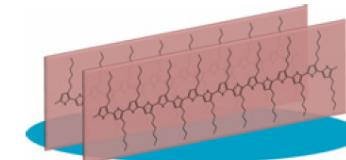
$$I(\Theta) = \int_{\text{peak}} PEY(\Theta)$$

Dichroic Ratio:

$$R = \frac{I(90^\circ) - I(0^\circ)}{I(90^\circ) + I(0^\circ)}$$

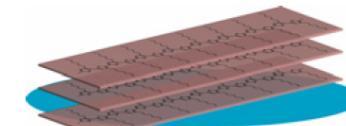
for conjugated plane

Edge-On



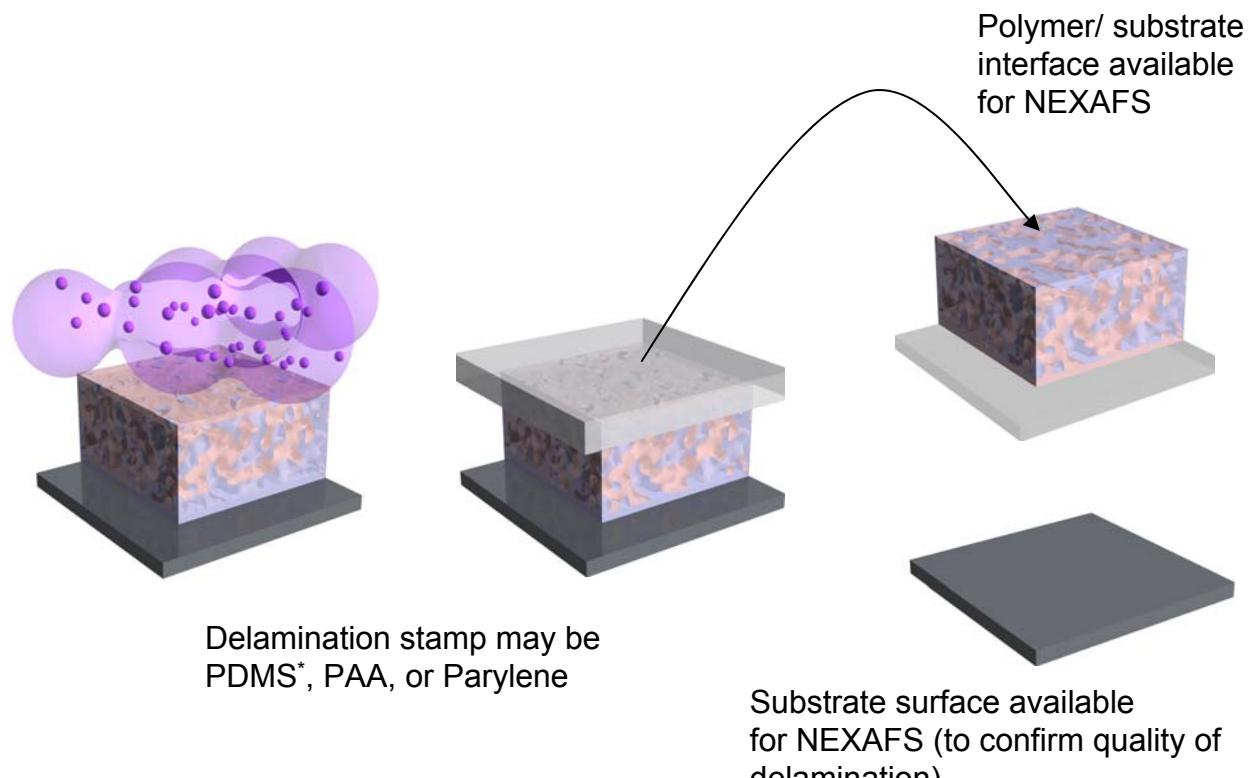
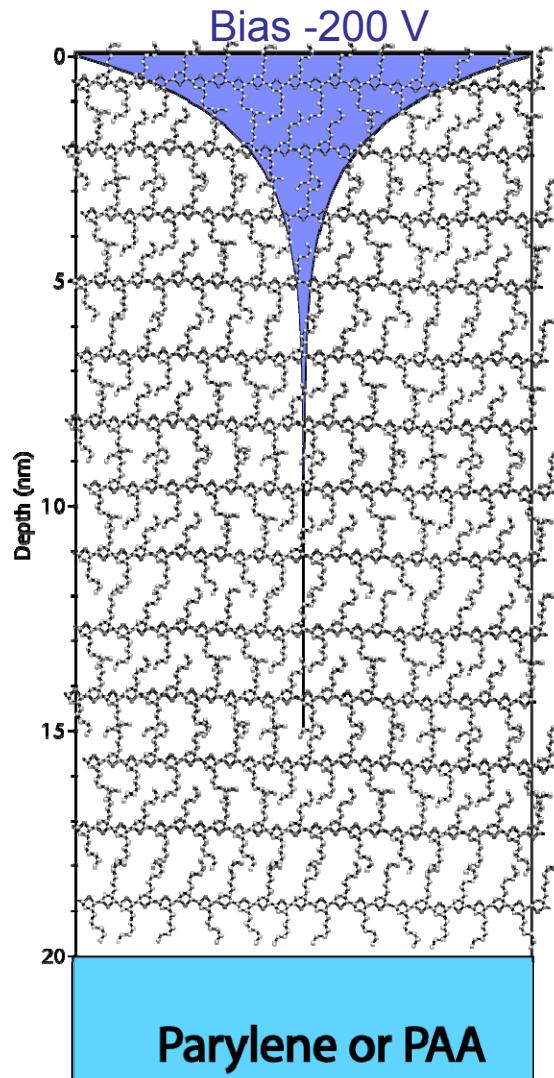
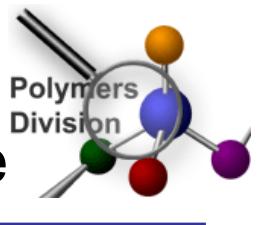
$$R = 0.70$$

Plane-On



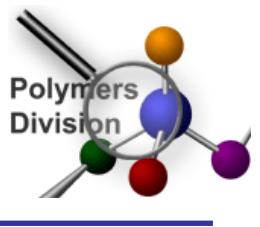
$$R = -1.00$$

NEXAFS Probes Surfaces: Delamination Exposes the Polymer/ Substrate Interface

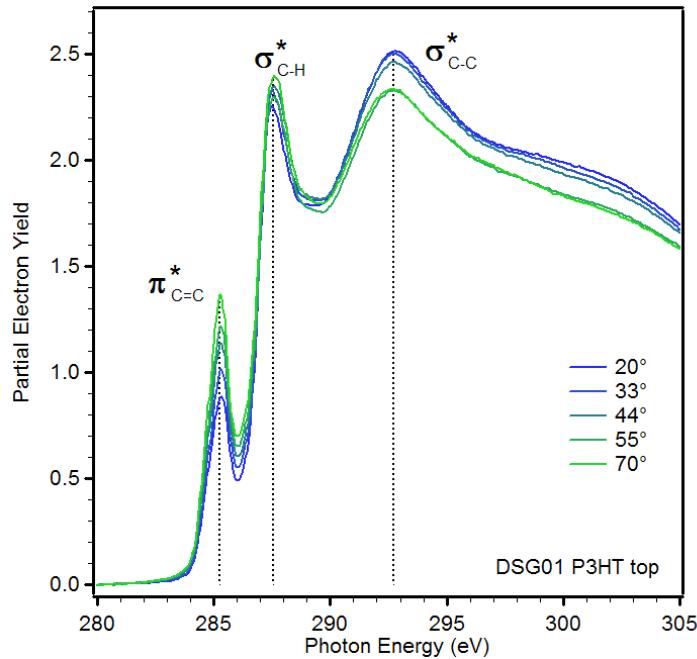


* Chabiny et al. JACS 2004, 126 (43), 13928-13929

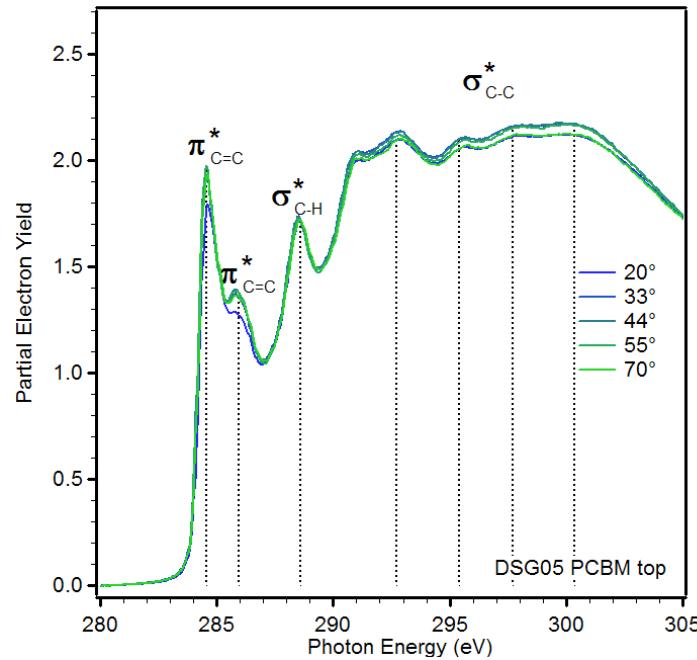
P3HT and PCBM Polymer/Air Interface Spectra



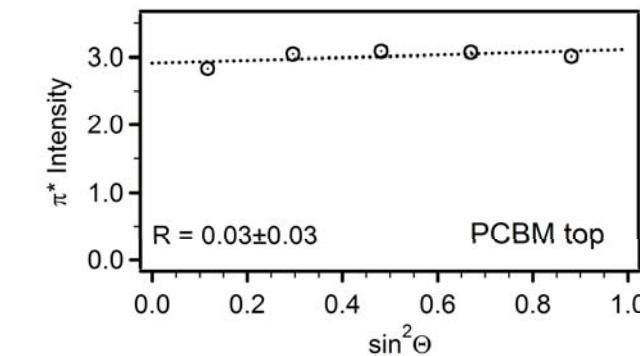
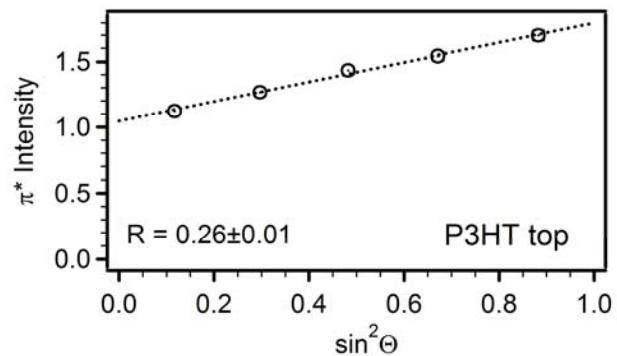
P3HT



PCBM

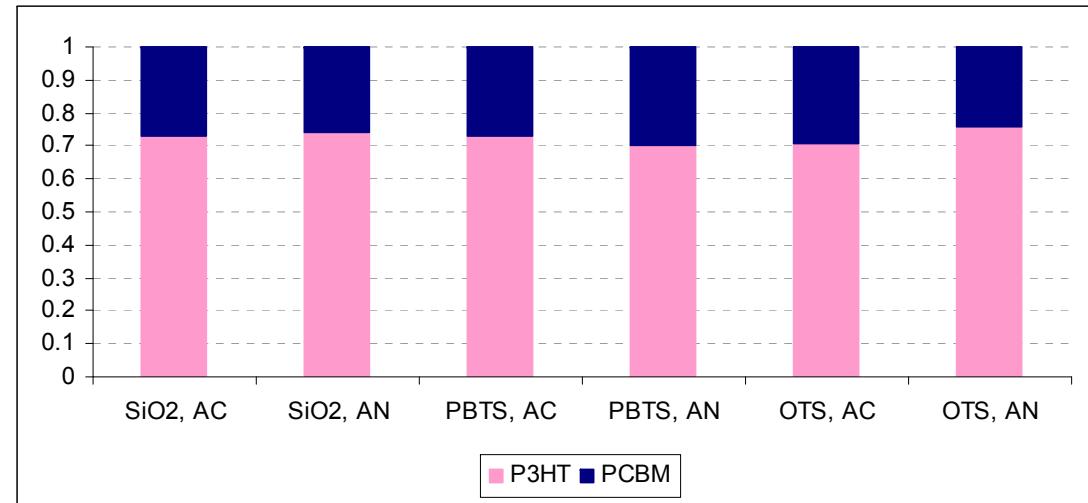
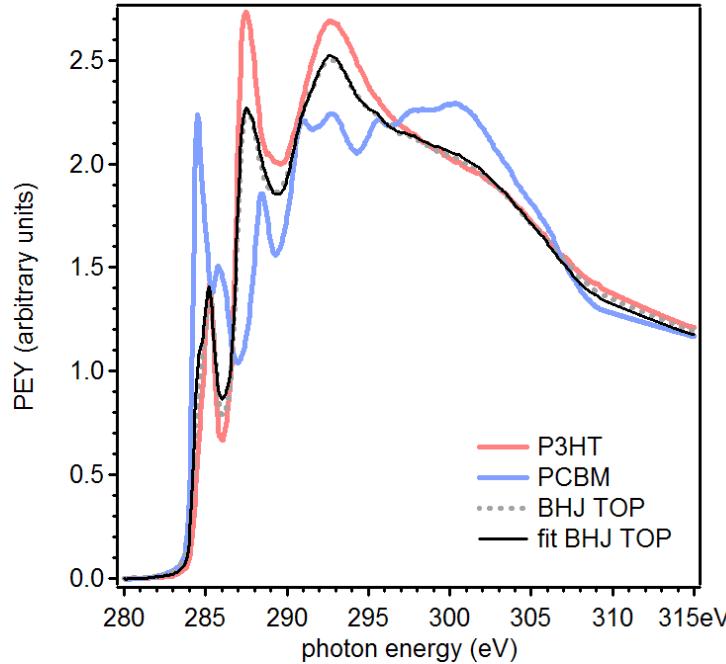
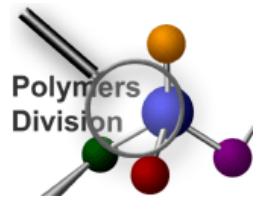


edge-on

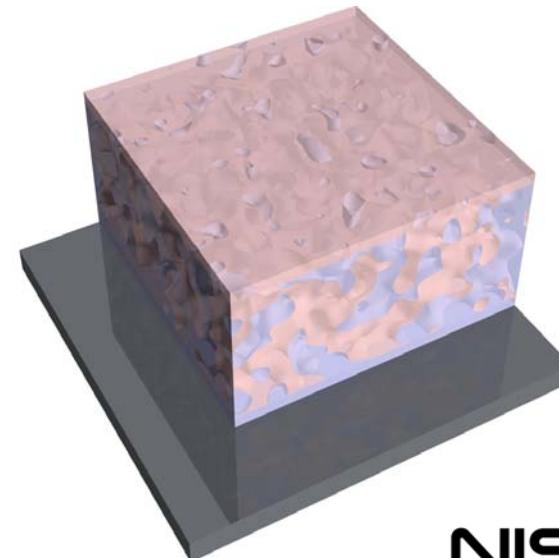


unoriented

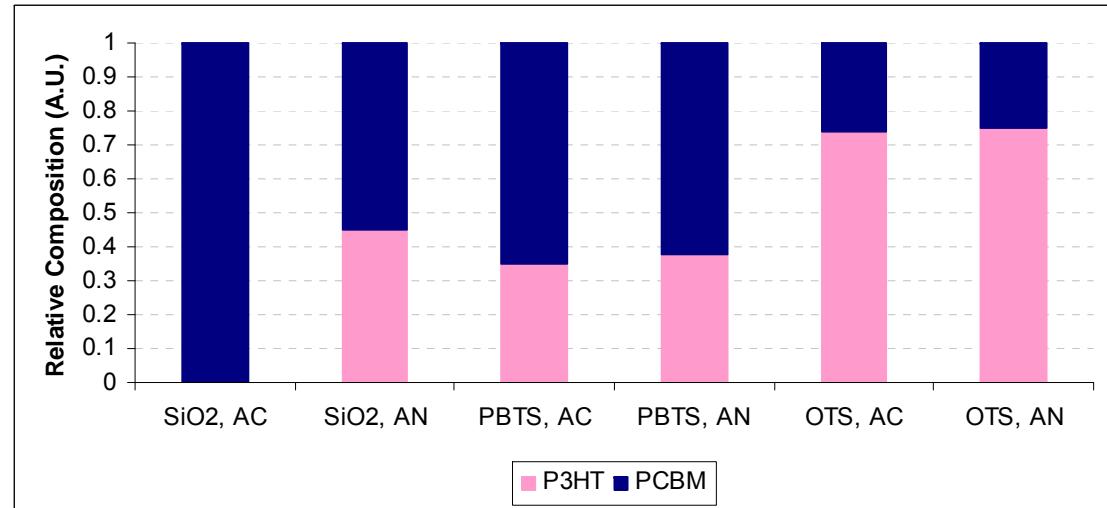
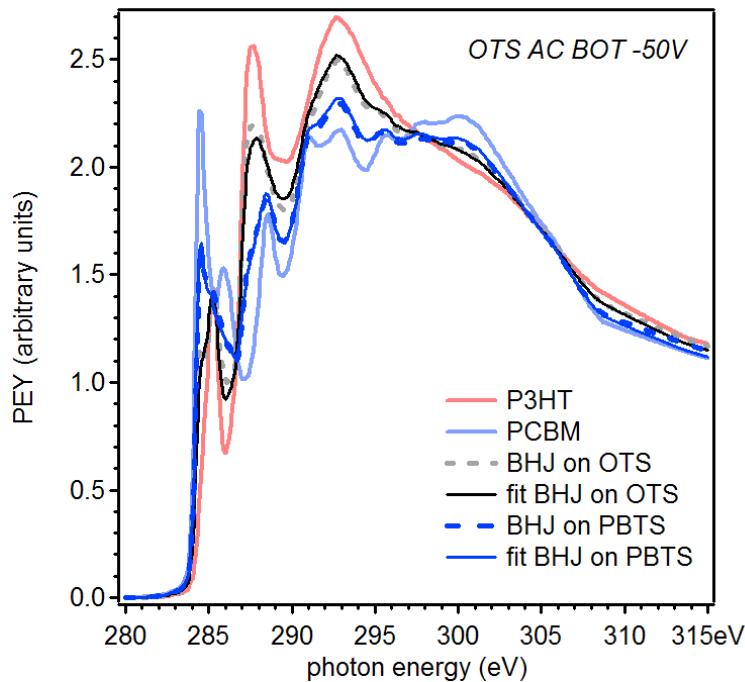
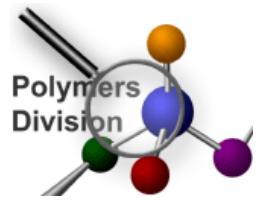
BHJ Polymer/ Air Surface Spectra: P3HT Enrichment Under All Circumstances



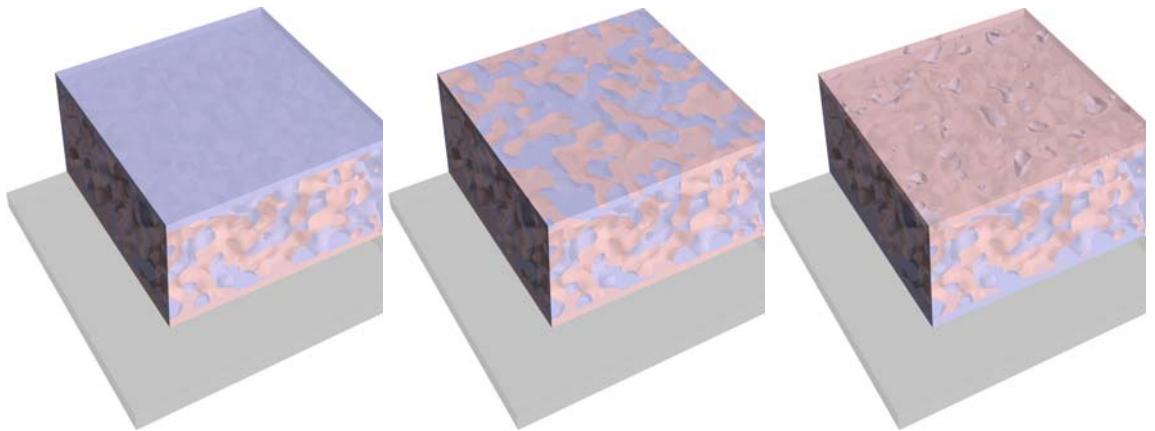
Subs.	Sample	P3HT	PCBM	R
SiO ₂	BHJ AC	0.73	0.27	0.12 ± 0.03
SiO ₂	BHJ AN	0.74	0.26	0.09 ± 0.01
PBTS	BHJ AC	0.73	0.27	0.26 ± 0.01
PBTS	BHJ AN	0.70	0.30	0.09 ± 0.02
OTS	BHJ AC	0.71	0.29	0.16 ± 0.02
OTS	BHJ AN	0.76	0.24	0.13 ± 0.05



BHJ Polymer/ Substrate Interface Spectra: Composition Changes with Surface Energy (γ)

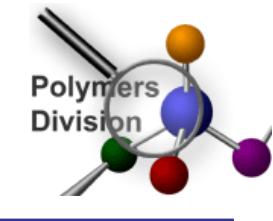


Subs.	Sample	P3HT	PCBM	R
SiO ₂	BHJ AC	0.00	1.00	0.02 0.01
SiO ₂	BHJ AN	0.45	0.55	0.05 0.01
PBTS	BHJ AC	0.35	0.65	0.10 ± 0.03
PBTS	BHJ AN	0.38	0.62	0.01 ± 0.00
OTS	BHJ AC	0.74	0.26	0.25 ± 0.03
OTS	BHJ AN	0.75	0.25	0.24 ± 0.00



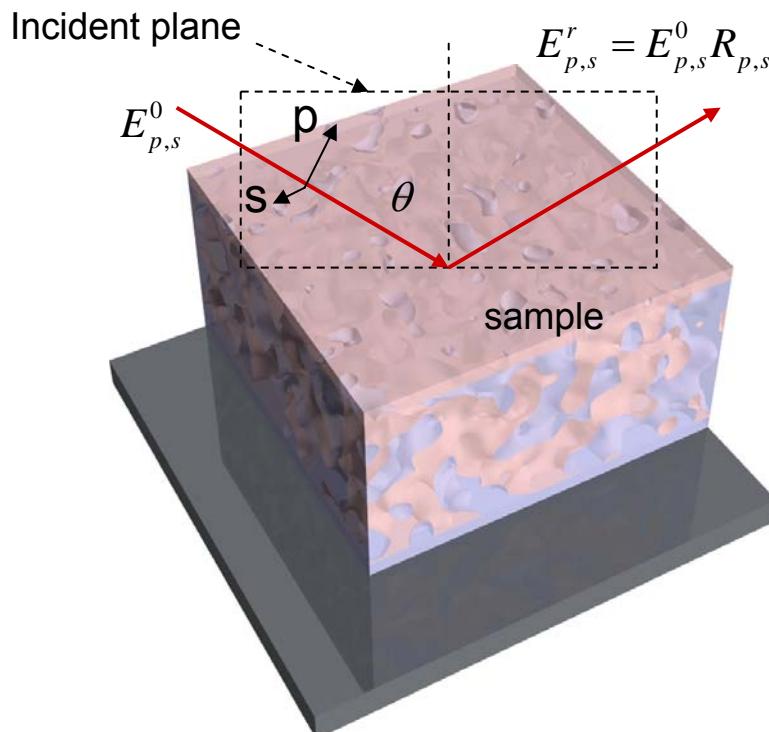
7, NOVEMBER 2008
DAVID S. GERMACK PH.D.

NIST Combinatorial Methods Center
Meeting No. 14



Variable Angle Spectroscopic Ellipsometry

In a VASE experiment a UV-Vis (or IR) spectrum and an ellipsometric measurement are simultaneously collected allowing for the analysis of samples with complex or unknown dielectric constants to be analyzed.

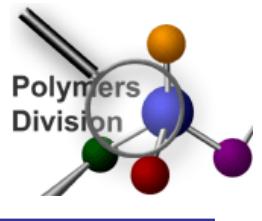


$$\rho \equiv \frac{R_p}{R_s} \equiv \tan \Psi e^{i\Delta}$$

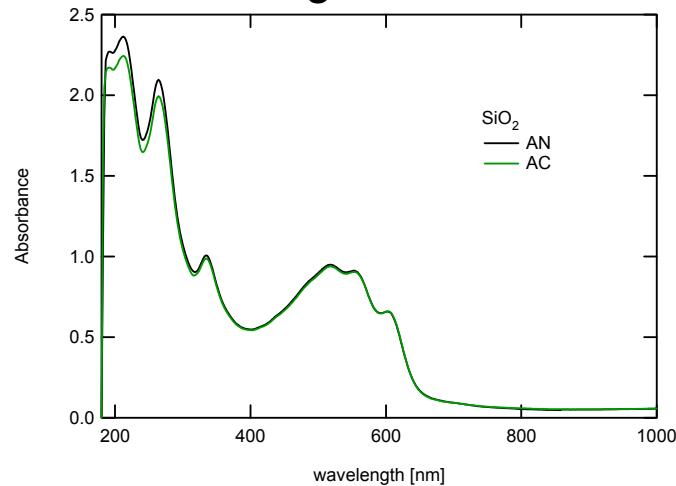
- VASE is model dependent
- complimentary optical techniques inform the model and provide checks on the “physicality” of the model
- models can include materials properties like optical anisotropy and conductivity

Example of VASE applied to BHJs:
Campo-Quiles, M. et al. *Nature Materials* 2008, 7, 158-164

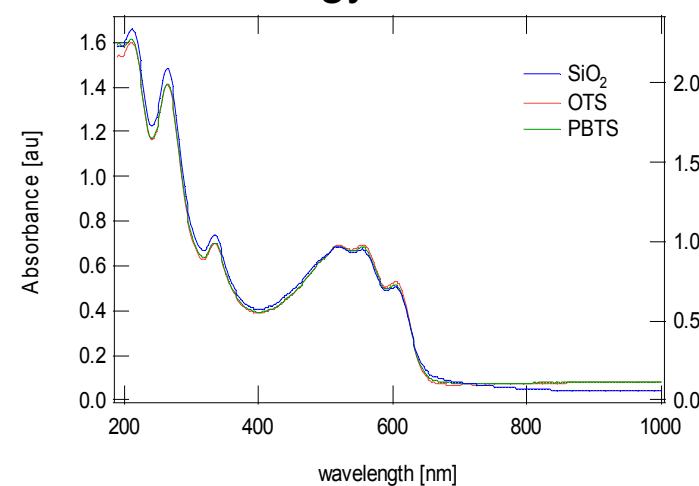
Before Constructing a VASE Model it Helps to Know Some Things



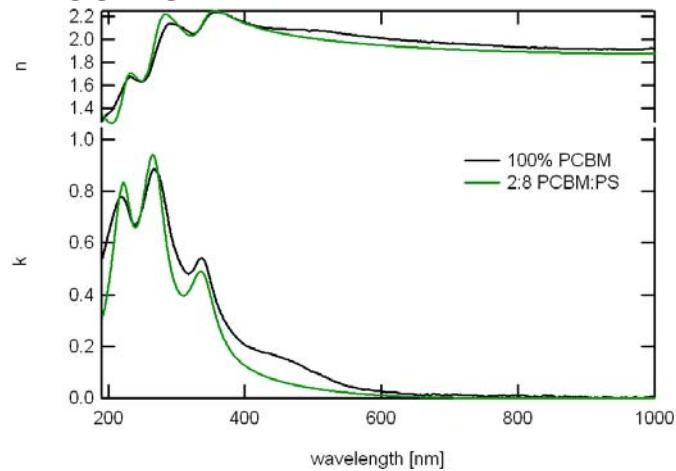
Thermal Budget: BHJ



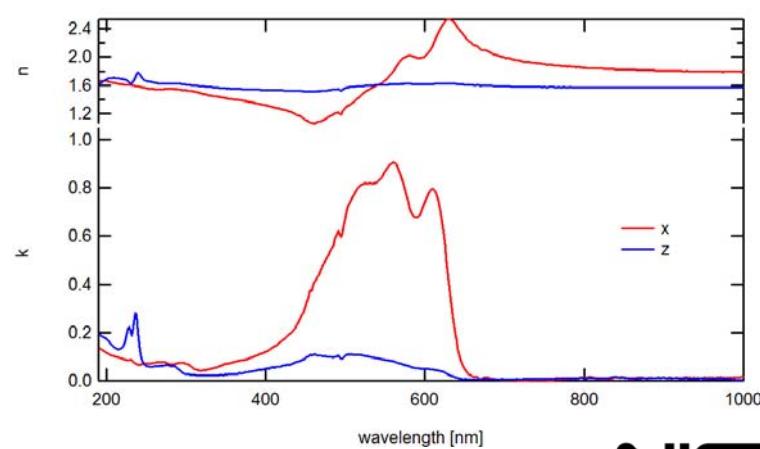
Surface Energy: BHJ



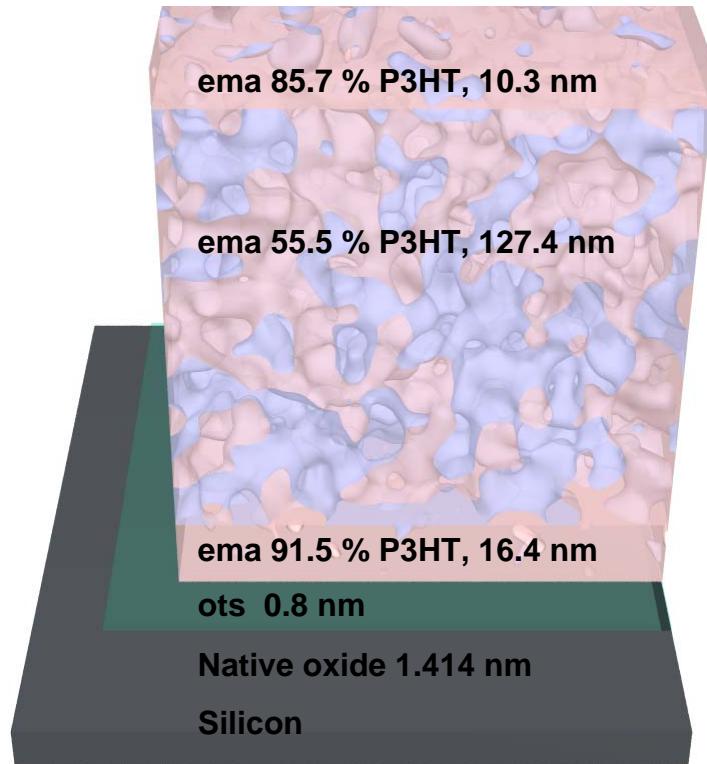
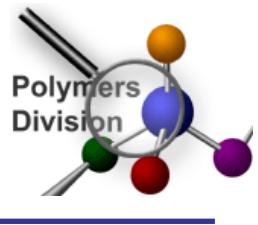
Aggregation: PCBM



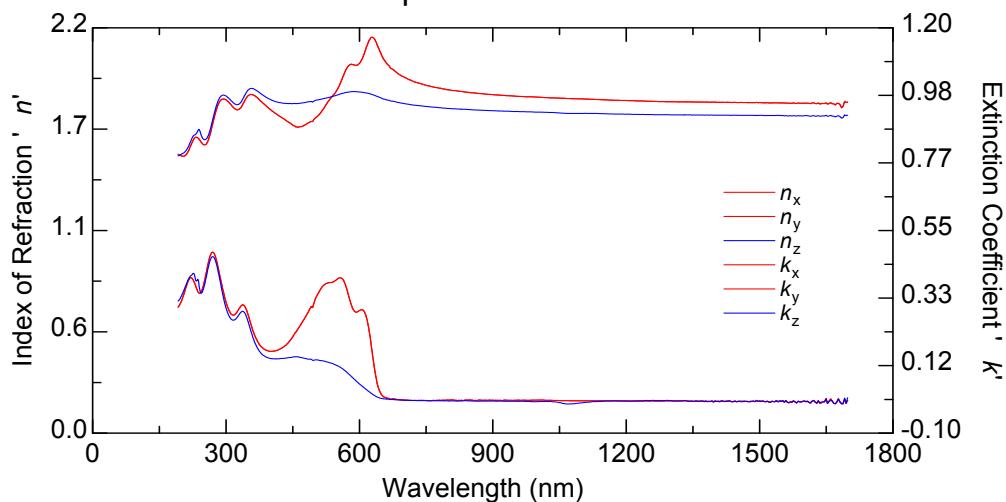
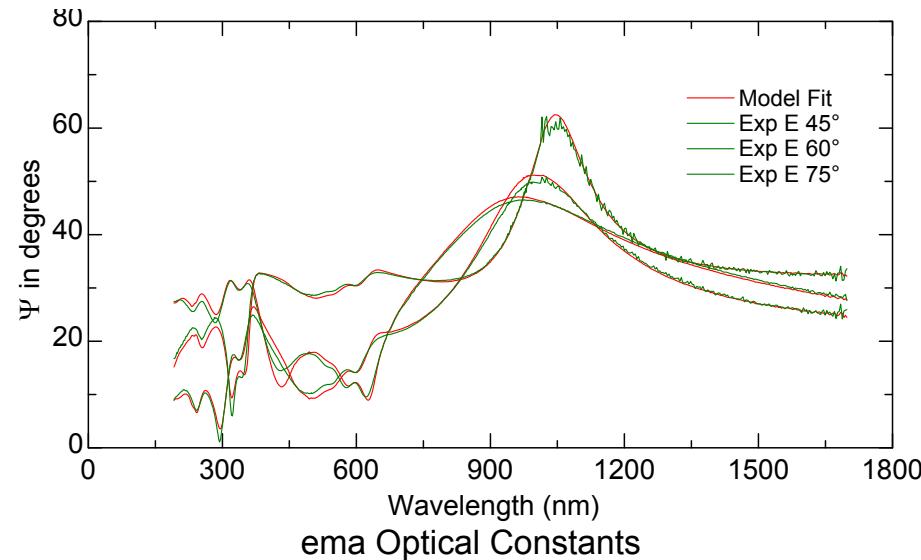
Orientation: P3HT



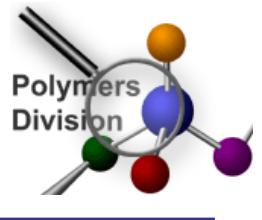
Raw VASE Data, Fitted Curves and Model



ema = *effective medium approximation*
describes bulk properties based upon
the “mixing” of its constituents, in this case
PCBM and P3HT



Optimized VASE Model: Good Agreement with NEXAFS



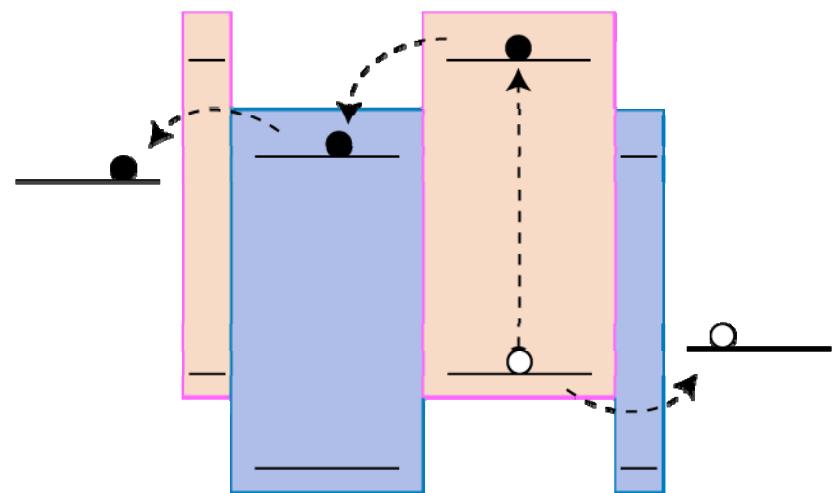
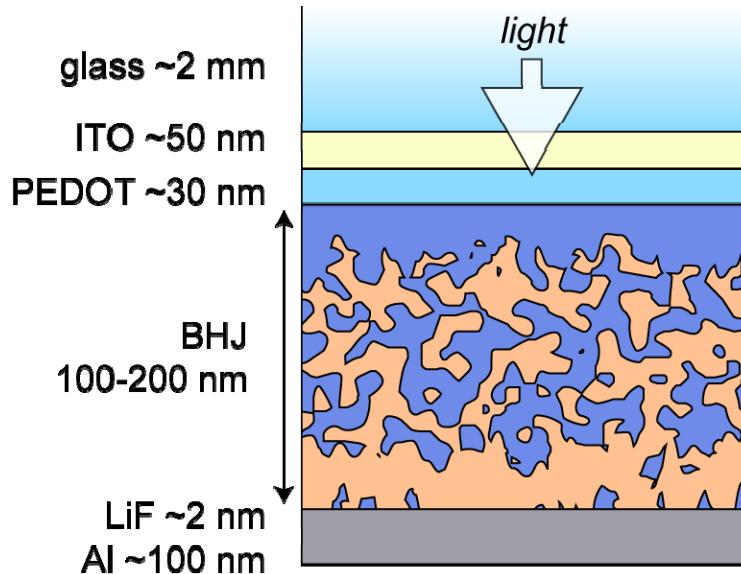
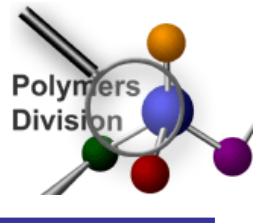
	Top				Middle			Bottom		
	γ (mN/m ²)	% P3HT _N	% P3HT _{SE}	t _{SE} (nm)	% P3HT _{SE}	$\varepsilon_x : \varepsilon_z$	t _{SE} (nm)	% P3HT _N	% P3HT _{SE}	t _{SE} (nm)
SiO ₂	77.36*	74	100	10	59	4:1	68	0	28	8
PBTS	26.23*	74	83	13	56	3:1	132	35	100	7
OTS	20.03*	75	86	10	56	4:1	127	74	92	16
P3HT	26.9†	-	-	-	-	-	-	-	-	-
PCBM	37.8‡	-	-	-	-	-	-	-	-	-

* Janssen, D. et. al. *Thin Solid Films* **2006**, 515, 1433-1438

† Wang, X.; Ederth, T.; Inganäs, O., *Langmuir* **2006**, 22, 9287-9294

‡ Nilsson, S.; Bernasik, A.; Budkowski, A.; Moons, E., *Macromolecules* **2007**, 40 8291-8301

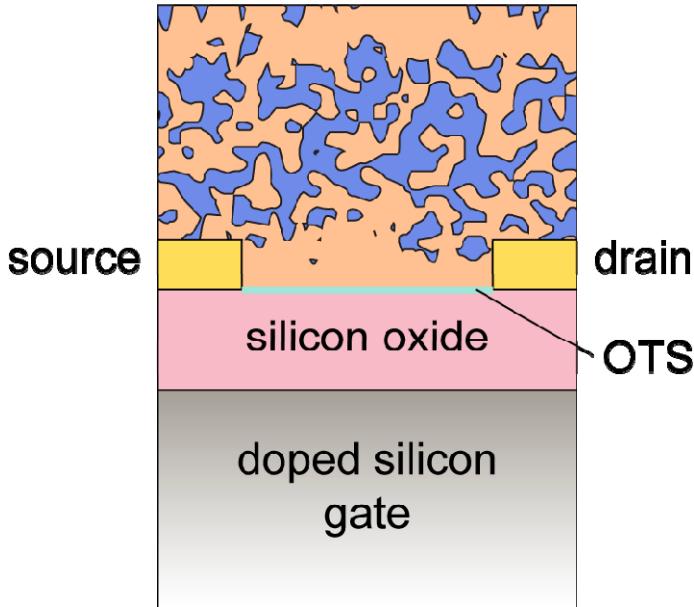
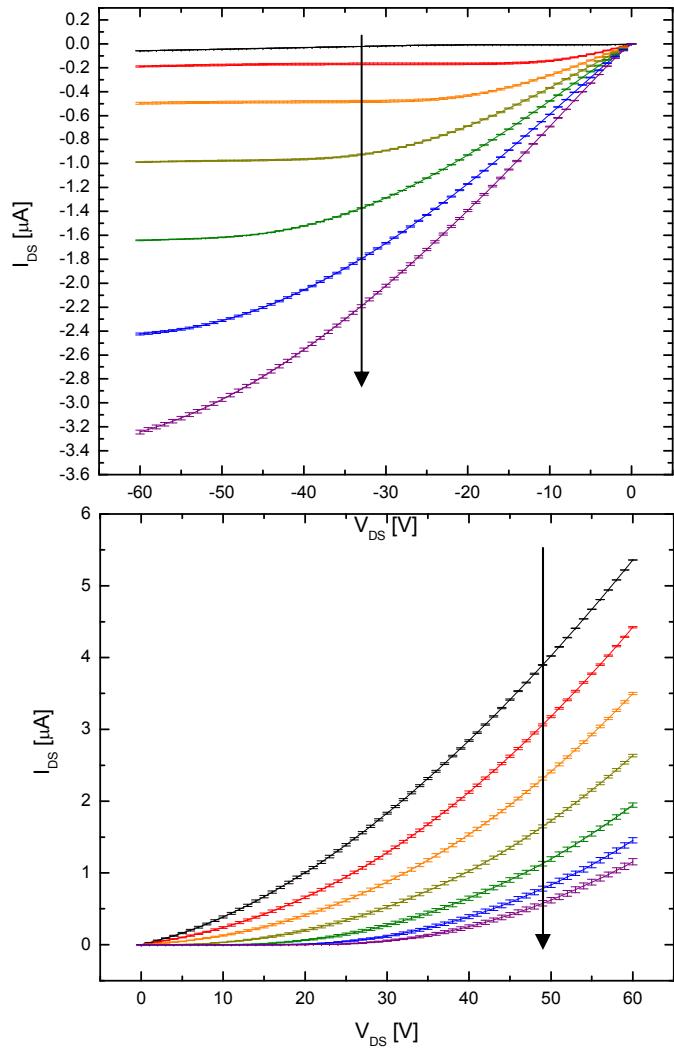
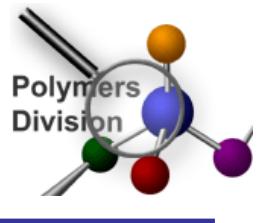
Vertical Phase Distribution Atop PEDOT:PSS



Potential cause of poor ff?

- But, blocking layers are very thin relative to BHJ (each is 5-10% BHJ thickness)
- P3HT layer might be removed or changed by metal evaporation
- Blocking layers are not “perfect,” so lateral percolation might allow carriers to go around them (at the cost of some additional series resistance).
- Suggests that surface energy of HIL might be an important optimization variable for more reasons than simple solvent wetting.

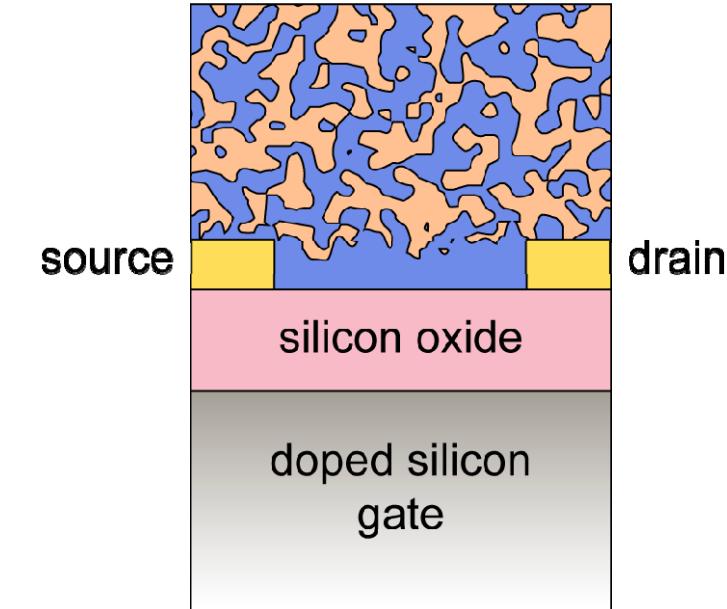
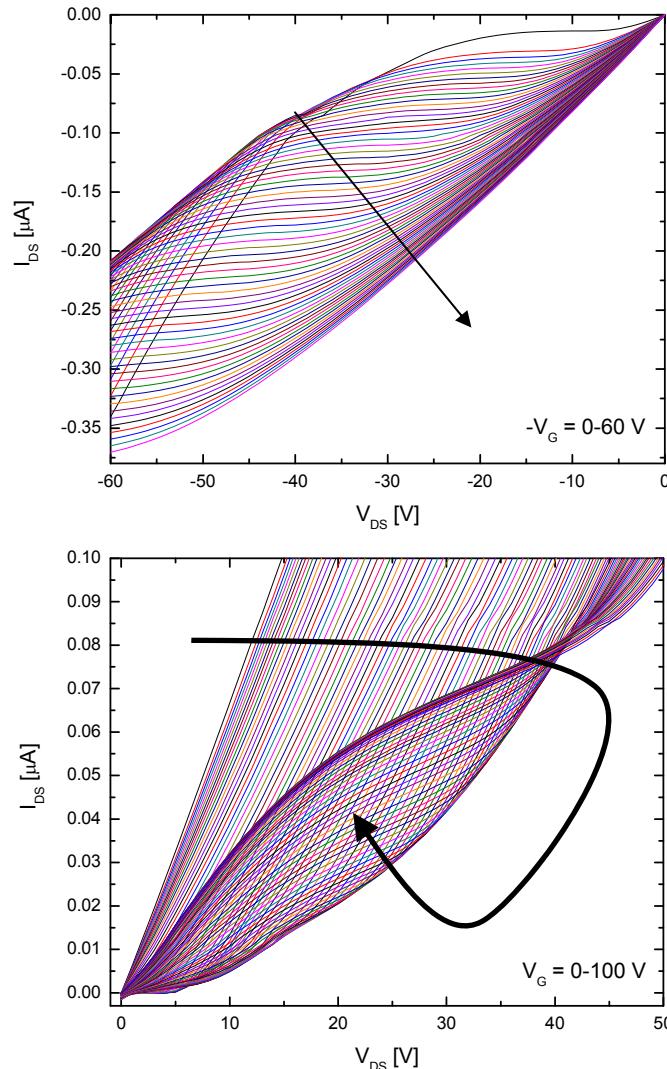
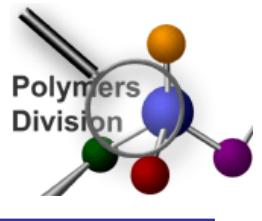
OTFTs of Bulk Heterojunctions: Dependence on Dielectric Surface Energy



On OTS

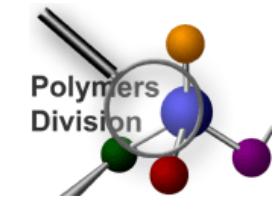
- P3HT-rich bottom layer
- Only observe hole transport
- $\mu_h = 1.75 \times 10^{-3} \text{ cm}^2\text{V}^{-1}\text{s}^{-1}$

OTFTs of Bulk Heterojunctions: Dependence on Dielectric Surface Energy



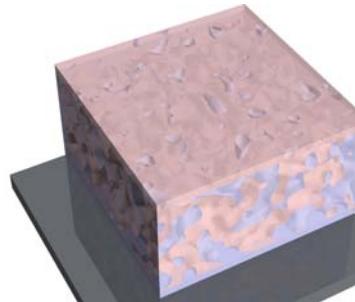
On bare oxide

- PCBM-rich bottom layer
- Observe ambipolar transport
- Difficult to extract hole and electron mobilities independently.



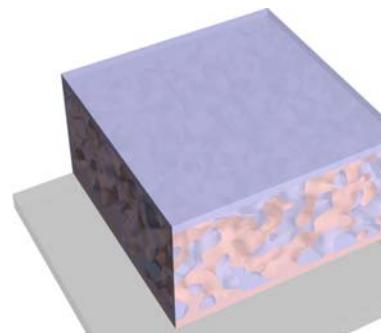
Summary

Polymer/ Air Interface:
aka Cathode

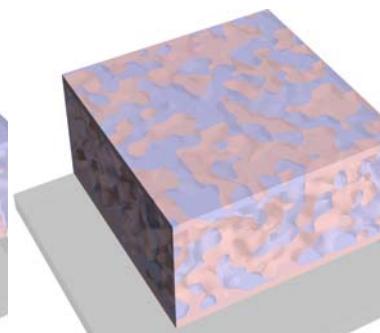


All Cases

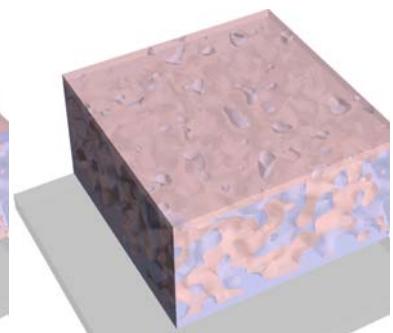
Polymer/ Substrate Interface:
aka Anode



As Cast Oxide



Annealed Oxide/
PBTS

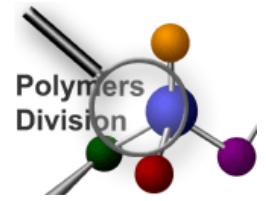


OTS

TFT – like orientation of P3HT chains throughout the BHJ is non-ideal for vertical transport of holes

Future Work

- Interrogate BHJ layers on HTL *via* VASE
- Apply knowledge gained here to OPV devices by assembling SAMs on HTL surfaces
- Explore relationship of Anode/BHJ and Cathode/BHJ interfaces and device aging
- Correlate VASE model to reflectivity/ diffraction/ scattering results



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